

University of Kerala

Discipline	PHYSICS								
Course Code	UK3DSEPHY202								
Course Title	BASICS OF	NANOSCIEN	CE AND NAN	OTECHNOLO	CY				
Type of Course	DSE			1	· (
Semester	Ш			M					
Academic Level	200 - 299			U_{k}					
Course Details	Credit	Lecture per week	Tutorial per week	Practical per week	Total Hours/Week				
	4	4 Hrs	7.7	-	4 Hrs				
Pre-requisites	-)						
Course Summary	Electrical an		erties of Materia	als, Generation	of Nanoscience otechnology for				

BOOKS FOR STUDY:

- 1. Materials Science and Engineering: An Introduction, 10th Edition by William D. Callister Jr, David G. Rethwise, Wiley (2018)
- 2. Textbook of Nanoscience And Nanotechnology, B.S Murthy, P. Shankar. Baldev Raj, B.B.Rath, James Murday, Orient Blackswan, (2021)
- 3. Nanotechnology: Principles and Practices, Third Edition, by Sulabha K. Kulkarn (2014)
- 4. Introduction To Nanoscience And Nanotechnology By Chattopadhyay, PHI, India
- 5. Introduction to Nanotechnology, Charles P. Poole Jr., Frank J. Owens, Wiley-Interscience

BOOKS FOR REFERENCE:

- Fundamentals of Nanotechnology, CRC press, by G.L. Hornyak, J.J. Moone, H.F. Tihhale, J. Dutta
- 2. Nano Essentials- T.Pradeep, TMH
- 3. Nanomaterials by A.K. Bandyopadhyay; New Age International Publishers.
- 4. Nanotechnology by Mark Ratner and Daniel Ratner, Pearson Education.

WEB REFERENCE

1. https://nptel.ac.in/courses/113/104/113104076/

DETAILED SYLLABUS: THEORY

Module	Unit	Content	firs	CO No
		Materials Structure and Bonding	12	
	1	Why Study Materials Science and Engineering? (Book1: Chapter 1)	1	1
	2	Crystal Structure- Fundamental Concepts, Unit Cell, *Elementary ideas and examples of Metallic Crystal structures (Book1: Chapter 3)	2	1
I	3	Elementary ideas and examples of Three-Dimensional Crystal systems, Planes in the Crystals, Crystallographic Directions, (Book1: Chapter 3)	4	1
	4	Crystalline and non-crystalline materials Single crystals, Polycrystals (Book1: Chapter 3)	2	2,3
	5	Atomic packing factors of FCC, BCC, Hexagonal close packed crystal structure (Book1: Chapter 3)	3	2,3
	Y	Inter-atomic bonding	8	
	6	Atomic bonding in solids - bonding forces and energies (Book1: Chapter 2)	3	1
II	7	Primary Interatomic bonds - Ionic bonding, Covalent bonding (Book1: Chapter 2)	3	1
	8	Metallic bonding, Secondary bonding (van der Waals bonding) (Book1: Chapter 2)	2	1
		Electrical and Optical Properties of Materials	16	
III	9	Electrical Conductivity, Electronic and Ionic Conduction (Book1: Chapter 18)	3	4

			1	1
	10	Energy Band Structures in Solids, Conduction in Terms of Band and Atomic Bonding Models, Electron Mobility (Book1: Chapter 18)	4	4
	11	Semiconductivity- Intrinsic Semiconduction, Extrinsic Semiconduction (n-Type Extrinsic Semiconduction & p-Type Extrinsic Semiconduction) (Book1: Chapter 18)	4	4
	12	Electromagnetic Radiation, Light Interactions with Solids, Atomic and Electronic Interactions (Book1: Chapter 21)	2	4
	13	Optical properties of Non metals: Refraction, Reflection, Absorption, Transmission, Colour (Book1: Chapter 21)	3	4
		Generation of Nanoscience and Nanotechnology	12	
	14	Nano – History and Scope, Cluster and Magic Numbers, Early applications of nanotechnology -Nano-gold (Book 2: Chapter 1)	2	4
	15	Applications of Nanomaterials, Nature: The Best Nanotechnologist - web-spinning spider, lotus-leaf effect, Water striders (Book 2: Chapter 1)	2	4
IV	16	Effect of nano-dimensions on materials behaviour: Elastic properties & Melting point (Book 2: Chapter 2)	2	4,5
	17	Clusters- Types of Clusters, Semiconductor nanoparticles, Excitons (Book 3: Chapter 8)	3	4,5
	18	The Era of new nanostructures of Carbon – Buckminsterfullerene, Carbon Nanotubes (CNTs), Multi walled CNTs (Book 4: Chapter 8, Book 5: Chapter 5 Section 5.1 & 5.2 ONLY)	3	5
	A	pplications of Nanoscience and Nanotechnology for	12	
V*	19	a Sustainable Future Can small things make a big difference? Nanoscience and Properties, Classification of nanostructured materials (Book 2: Chapter 1, Book 5: Chapter 4)	4	4
	20	Fascinating nanostructures (Book 2: Chapter 1)	4	5
	21	Nanotechnology and Environment (Book 3, Chapter 11)	4	6

COURSE OUTCOMES

CO No.	Upon completion of the course the graduate will be able to	Cognitive Level	PSO addressed
CO-1	Explain the fundamentals of nanoscale systems and its physical, chemical and electrical properties	U	PSO-1,2
CO-2	List and classify the fundamental crystal structure of materials	Ap	PSO-2,3
CO-3	Outline the various bonds occurs in materials	U	PSO-1,3
CO-4	Formulate the concepts of electrical and optical properties of materials and compare it with that of nanomaterials	S	PSO-3,4
CO-5	Design models to demonstrate the influence of surface effects on the mechanical, physical, and chemical properties of materials such as CNTs and MWCNTs	, c	PSO-5
CO-6	Analyse the fundamental applications of nanotechnology and point out how it supports for a sustainable future in modern era	An	PSO-6,7

R-Remember, U-Understand, Ap-Apply, An-Analyse, E-Evaluate, C-Create

Name of the Course: BASICS OF NANOSCIENCE AND NANOTECHNOLOGY

Credits: 4:0:0 (Lecture: Tutorial: Practical)

CO No.	СО	PO/PSO	Cognitive Level	Knowledge Category	Lecture (L)/ Tutorial (T)	Practical (P)
CO-1	Explain the fundamentals of nanoscale systems and its physical, chemical and electrical properties	PO 1,3,4,5,6, 8/ PSO-1,2	U	F, C	L	-
CO-2	List and classify the fundamental crystal structure of materials	PO 1,2,3,4,5, 8/ PSO-2,3	Ap	F, C	L	-
CO-3	Outline the various	РО	U	F, C	L	-

	bonds occurs in	1,2,3,4,6/				
	materials	PSO-1,3				
	Formulate the					
	concepts of					
	electrical and optical	PO				
CO-4	properties of	1,2,3,4,6/	C	F, C	L	-
	materials and	PSO-3,4				
	compare it with that					
	of nanomaterials					
	Design models to					
	demonstrate the				JV	
	influence of surface					
	effects on the	PO			X	
CO-5	mechanical,	1,2,3,7/	C	F, C	L	-
	physical, and	PSO-5				
	chemical properties					
	of materials such as					
	CNTs and MWCNTs					
	Analyse the					
	fundamental					
	applications of	PO				
CO-6	nanotechnology and	1,3,4,5,7,	An	F, C	L	
0-0	point out how it	8/	All	Γ, C	L	-
	supports for a	PSO-6,7				
	sustainable future in					
	modern era					

F-Factual, C- Conceptual, P-Procedural, M-Metacognitive

Mapping of COs with PSOs and POs:

	P S O 1	P S O 2	P S O 3	P S O 4	P S O 5	P S O 6	P S O 7	P O 1	P O 2	P O 3	P O 4	P O 5	P O 6	P O 7	P O 8
CO-1	2	2	-	-	-	-	-	2	-	1	1	-	2	-	1

CO-2	-	2	1	-	-	-	-	1	2	2	1	-	-	-	2
CO-3	2	-	1	-	-	-	-	1	2	2	1	-	1	-	-
CO-4	-	-	3	2	-	-	-	2	2	1	1	-	1	-	-
CO-5	-	-	-	-	3	-	-	1	2	2	-	-	-	2	-
CO-6	-	-	-	-	-	2	2	1	-	2	2	-	-	1	1

Correlation Levels:

Level	-	1	2	3
Correlation	Nil	Slightly / Low	Moderate / Medium	Substantial / High

Assessment Rubrics:

- Quiz / Assignment/ Discussion / Seminar
- Midterm Exam
- Programming Assignments
- Final Exam

Mapping of COs to Assessment Rubrics

CO No	Internal Exam	Assignment	Project Evaluation	End Semester Examinations
CO-1	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	-	-	✓
CO-2		-	-	✓
CO-3	1	✓	-	✓
CO-4	/	✓	-	✓
CO-5	1	√	-	✓
CO-6	✓	-	-	-



University of Kerala