

University of Kerala

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Discipline	PHYSICS							
Course Code	UK1DSCPHY102							
Course Title	PROPERTIES OF SOLIDS							
Type of Course	DSC							
Semester	Ι							
Academic Level	100 - 199							
Course Details	Credit	Lecture per week	Tutorial per week	Practical per week	Total Hours/Week			
	4	3 Hrs	-	2 Hrs	5 Hrs			
Pre-requisites	 Students sh chemical b Students sh ferro magn Students sh expansion Students sh electronics 	 Students should have a basic understanding of states of matter and chemical bonding. Students should be aware of basics of diamagnetic, paramagnetic, and ferro magnetic substances. Students should know the fundamentals of heat, temperature, thermal expansion and heat transfer by conduction, convection and radiation. Students should be familiar with foundation of semiconductor electronics 						
Course Summary Course								

BOOKS FOR STUDY:

Physics, Principles with Applications, Douglas C. Giancoli, Pearson Education Limited, 7th Engineering Physics, G Aruldhas, PHI Learning Private Limited, New Delhi(2012).

3. Heat and Thermodynamics: Brijlal and Subramaniam, S. Chand &Co.

BOOKS FOR REFERENCE:

- Concepts of Modern Physics, Arthur Beiser, Shobhit Mahajan, S Rai Choudhury, McGraw Hill Education (India) Private Limited (2017).
- 2. Solid State physics and Electronics, R K Puri, V K Babbar, S Chand & Company Ltd, (2008).
- 3. Solid State Physics, Dr R Asokamani, Eswar Press, Chennai, (2015).
- 4. Applied Solid State Physics, Rajnikant, Wiely India Pvt. Ltd. 92011)
- 5. Heat and Thermodynamics: M. Zeemansky, McGraw Hill, New Delhi (2007).
- Heat and Thermodynamics: Rose C. McCarthy, The Rosen Publishing Group, Inc. NY,(2005)
 - 7. Thermodynamics Kinetic Theory and Statistical Thermodynamics: F. W. Sears and G.
 - 8. L. Salinger, Addison-Wesley Publishing Company, 3rd Edn. (1975).

DETAILED SYLLABUS: THEORY

Module	Unit	Content	Hrs	CO No
		MOLECULES AND SOLIDS (Book 1)	8	
	Bonding in molecules, Potential Energy diagrams for 1 molecules, Weak (Van der Waals) bonds – Protein synthesis, 2 Bonding in solids		2	1
I			1	1
	3	3 Free electron theory of metals		1
	4	Band theory of solids, Semiconductors and doping, Semiconductor Diodes, LEDs, OLEDs, Transistors: Bipolar and MOSFETs, Integrated Circuits, 22-nm Technology	3	1,6
		10		
п	5	Permeability and Susceptibility	1	2
	6	1	1,6	
	7	7 Classification of magnetic materials		

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	8	Curie Temperature, Weiss theory of Ferromagnetism	1	2			
	9	Hysteresis loop	2	2			
	10	Antiferromagnetism, Ferrimagnetism, Ferrites, Magnetic					
		materials for information storage	3	2			
		SUPERCONDUCTIVITY	Q				
		(Book 2)	0				
	11	11 Introduction, Magnetic Properties, Meissner Effect					
Ш	12	Type I and type II superconductors - London Penetration					
		depth		3			
	13	Isotope Effect, BCS Theory, Cooper Pairs	2	3			
-	14	High temperature superconductivity, Applications	2	3			
		THERMAL PROPERTIES	10				
		(Book 3)	10				
	15	Coefficient of Thermal Conductivity, Thermometric	1	4			
		Conductivity					
	16 16 flow		2	4,6			
IV	17	Wiedemann – Franz Law	1	4			
	18	Latent Heat of Fusion	1	4			
	19	Laws of Fusion - Practical Applications	1	4			
	20	1	4				
	21	Thermoelectric effects - Seebeck, Peltier, Thomson effect	3	4			
		and Thermo electric power		•			
		DIELECTRIC PROPERTIES	9				
		(Book 2)					
	22	Electric Dipole, Polarizability, Polarisation vector,	2	5			
V*		Dielectric constant, Dielectric Susceptibility					
		Dielectric polarisation- Electronic Folarisation, Ionic	2	E			
	23	Polarisation, Orentation Polarisation	5	5			
			Polarization, Iotal Polarisation				

24	Frequency dependence of Polarisation, Dielectric loss	2	5
25	Ferroelectric Crystals, Piezoelectric Crystals- Applications	1	5
26	Applications of dielectric materials	1	5

DETAILED SYLLABUS: PRACTICALS

Part A – At least 5 Experiments to be performed					
l No	0 Name of Experiment				
1	Diode Characteristics (for Ge and Si diodes)	6			
	Zener diode characteristics: To (i) trace and construct the circuit (ii) to plot				
2	the V-I characteristicunder reverse biased condition and (iii) to calculate the	6			
_	dynamic resistance of the diode under reverse bias when conducting				
3	Phase transition-determination of M.P of wax	6			
4	Determination of thermal conductivity of rubber	6			
5	Lee's disc-determination of thermal conductivity of a bad conductor	6			
6	Circular coil- magnetization of a magnet	6			
	Absolute determination of m and Bh using box type and Searle's type	6			
7	vibrationmagnetometers				
8	Searle's vibration magnetometer-comparison of magnetic moments	6			
9	Potentiometer – Resistivity	6			
	Part B* – At least One Experiment to be performed				
10	Thermo emf-measurement of emf using digital multimeter	6			
11	Determination of dielectric constant of a given material	6			
12	Carey Foster's Bridge-Resistivity	6			

COURSE OUTCOMES

No.	Upon completion of the course the graduate will be able to	Cognitive Level	PSO addressed
CO-1	Differentiate various types of bonding in solids, describe the relationship between band structure and electrical properties and apply the principles of doping in the fabrication of semiconductor devices	U, Ap	PSO-1,2
со-2	Illustrate and interpret the magnetic properties of materials and their theoretical underpinnings crucial for different applications	U, Ap	PSO-1,2
со-3	Explain the principles of superconductivity, describe the properties of superconducting materials, and evaluate the wide range of applications of superconductivity	U, Ap	PSO-1,2
CO-4	Describe and apply the thermal conductivity principles of solids, analyse the latent heat of fusion and evaluate its applications and analyse the thermoelectric properties of solids	U, Ap, An	PSO-1,2
CO-5	Describe the principles governing dielectric materials and categorise different types of dielectric crystals and appraise	U, Ap	PSO-1,2
СО-6	their application. Experiment and infer conductivity and magnetic property measurements as well as semiconductor device	Ap, An	PSO-1,2, 7
	characterization.	C-Create	

R-Remember, U-Understand, Ap-Apply, An-A

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Name of the Course: PROPERTIES OF SOLIDS

Credits: 3:0:1 (Lecture: Tutorial: Practical)

CO No.	со	PO / PSO	Cognitive Level	Knowledge Category	Lecture (L)/ Tutorial (T)	Practical (P)
CO-1	Differentiate various types of bonding in solids, describe the relationship between band structure and electrical properties and apply the principles of doping in the fabrication of semiconductor devices	PO 1, 2, 3, 4, 5, 8/ PSO – 1,2	U, Ap	С	L	-
CO-2	Illustrate and interpret the magnetic properties of materials and their theoretical underpinnings crucial for different applications	PO 1, 2, 3, 4, 6, 8/ PSO - 1,2	U, Ap	С	, L	-
CO-3	Explain the principles of superconductivity, describe the properties of superconducting materials, and evaluate the wide	PO 1, 2 3, 4, 8/ PSO – 1,2	, U, Ap	С	L	-

			The second se			
	range of applications of superconductivity					
CO-4	Describe and apply the thermal conductivity principles of solids, analyse the latent heat of fusion and evaluate its applications and analyse the thermoelectric properties of solids	PO 1, 2, 3, 4, 5, 8/ PSO – 1,2	U, Ap, An	С	L	-
CO-5	Describe the principles governing dielectric materials and categorise different types of dielectric crystals and appraise their application.	PO 1, 2, 3, 4, 8/ PSO – 1,2	U, Ap	С	-	-
CO-6	Experiment and infer conductivity and magnetic property measurements as well as semiconductor device characterization.	PO 1, 2, 3, 4, 5, 6, 8/ PSO – 1,2,7	Ap, An	Р	-	Р

conving of COs with PSOs and POs :

Aspendent and a second s															
	PSO 1	PSO 2	PSO 3	PSO 4	PSO 5	PSO 6	PSO 7	PO 1	PO 2	PO 3	PO 4	PO 5	PO 6	PO 7	PO 8
CO-1	3	2	-	-	-	-	-	3	3	2	1	1	-	-	1
CO-2	3	3	-	-	-	-	-	3	3	2	1	-	1	-	1
CO-3	3	3	-	-	-	-	-	3	3	3	1	-	-	-	
CO-4	3	3	-	-	-	-	-	3	3	3	1	1	-	-	
CO-5	3	3	-	-	-	-	-	3	3	3	2	-	-	-	
00.6	3	3	-	-	-	-	3	3	3	3	3	3	3	-	3
0-0															

alation Levels:

Correlation De				3
Level	_	1	2	5
Leve			Moderate /	Substantial / High
Correlation	Nil	Slightly / Low	Medium	Subsum
Corr				

Assessment Rubrics:

- Quiz / Assignment/ Discussion / Seminar .
- Midterm Exam .
- Programming Assignments .
- Final Exam .

f COs to Assessment Rubrics :

Mapping o	f COs to Assessme	nt Rubrics :		End Semester
	Internal Exam	Assignment	Project Evaluation	Examinations
CUNO	Internation		_	1
CO-1	\checkmark	-	-	√
СО-2	√	1		1
со-3	√		_	1
со-4	-	1	_	-
CO-5	✓	-	1	· -
CO-6	1	-		